

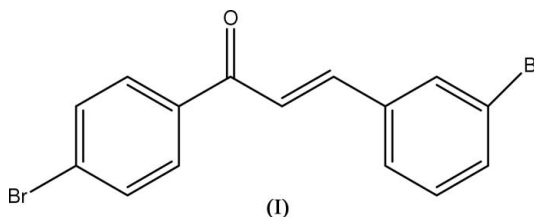
3-(3-Bromophenyl)-1-(4-bromophenyl)-  
prop-2-en-1-oneJeannie Bee-Jan Teh,<sup>a</sup> P. S. Patil,<sup>b</sup>  
Hoong-Kun Fun,<sup>a\*</sup>  
Ibrahim Abdul Razak<sup>a</sup> and  
S. M. Dharmaprasanth<sup>b</sup><sup>a</sup>X-ray Crystallography Unit, School of Physics,  
Universiti Sains Malaysia, 11800 USM, Penang,  
Malaysia, and <sup>b</sup>Department of Studies in  
Physics, Mangalore University, Mangalagan-  
gotri, Mangalore 574 199, India

Correspondence e-mail: hkfun@usm.my

## Key indicators

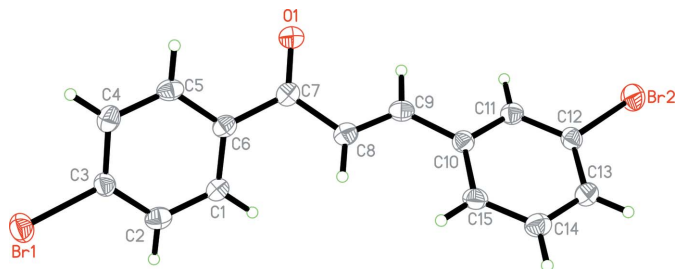
Single-crystal X-ray study  
T = 100 K  
Mean  $\sigma(\text{C}-\text{C}) = 0.006 \text{ \AA}$   
R factor = 0.047  
wR factor = 0.109  
Data-to-parameter ratio = 18.2For details of how these key indicators were  
automatically derived from the article, see  
<http://journals.iucr.org/e>.In the title compound,  $\text{C}_{15}\text{H}_{10}\text{Br}_2\text{O}$ , the dihedral angle  
between the benzene rings is  $48.0(1)^\circ$ . The crystal structure  
is stabilized by weak intermolecular  $\text{Br} \cdots \text{Br}$  contacts and  $\text{C}-$   
 $\text{H} \cdots \pi$  interactions.Received 4 May 2006  
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## Comment

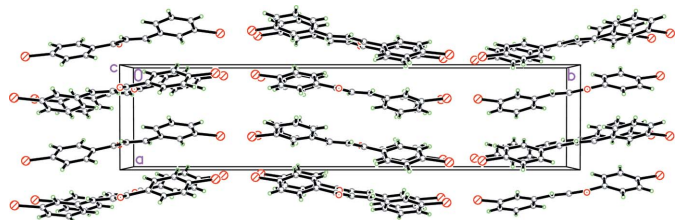
Chalcone derivatives are receiving increasing interest in the  
field of non-linear optics due to their excellent blue light  
transmittance, good crystal stability, large non-linear optical  
coefficients and relatively short cut-off wavelengths of trans-  
mittance (Fichou *et al.*, 1988; Kitaoka *et al.*, 1990; Uchida *et al.*,  
1998; Goto *et al.*, 1991; Patil *et al.*, 2006a,b; Zhang *et al.*, 1990;  
Zhao *et al.*, 2000). Here, we present the title compound, (I),  
crystals of which, however, do not exhibit second-order non-  
linear optical properties as they crystallize in a centrosym-  
metric space group.Bond lengths and angles in (I) display normal values (Allen  
*et al.*, 1987) and are comparable to those in related structures  
(Teh *et al.*, 2006; Patil *et al.*, 2006a,b; Ng *et al.*, 2006; Rosli *et al.*,  
2006). The least-squares plane through the enone fragment  
makes dihedral angles of  $26.6(2)$  and  $22.7(2)^\circ$  with the C1–C6  
and C10–C15 benzene rings, respectively. The dihedral angle  
between the benzene rings is  $48.0(1)^\circ$ .The relatively short distances between neighbouring Br  
atoms [ $\text{Br}1 \cdots \text{Br}2^i$   $3.7748(7) \text{ \AA}$ ,  $\text{Br}2 \cdots \text{Br}1^{ii}$   $3.7747(7) \text{ \AA}$ ,  
 $\text{Br}2 \cdots \text{Br}2^{iii}$   $3.7776(6) \text{ \AA}$  and  $\text{Br}2 \cdots \text{Br}2^{iv}$   $3.7776(6) \text{ \AA}$ ;  
symmetry codes: (i)  $2 - x, -\frac{1}{2} + y, \frac{3}{2} - z$ ; (ii)  $2 - x, \frac{1}{2} + y, \frac{3}{2} - z$ ;  
(iii)  $x, \frac{1}{2} - y, -\frac{1}{2} + z$ ; (iv)  $x, \frac{1}{2} - y, \frac{1}{2} + z$ ] indicate the presence of  
weak intermolecular  $\text{Br} \cdots \text{Br}$  interactions, which contribute to  
the stabilization of the crystal packing (Fig. 2), along with the  
 $\text{C}-\text{H} \cdots \pi$  interactions (Table 1).

## Experimental

Chalcone derivative (I) was obtained by the condensation of 3-  
bromobenzaldehyde (0.01 mol) with 4-bromoacetophenone  
(0.01 mol) in ethanol (60 ml) in the presence of NaOH (2 ml, 30%).  
After stirring for 2 h, the contents of the flask were poured into ice-  
cold water (250 ml), and left to stand for 24 h. The resulting crude  
solid was collected by filtration, dried and purified by repeated



**Figure 1**  
View of (I), with the atomic numbering and 50% probability displacement ellipsoids.



**Figure 2**  
The crystal packing, viewed down the *c* axis.

recrystallization from acetone. The purity of the compound was checked by thin-layer chromatography. Crystals suitable for single-crystal X-ray diffraction experiments were grown in 7 d by the slow evaporation of an acetone solution at room temperature.

#### Crystal data

$C_{15}H_{10}Br_2O$   
 $M_r = 366.05$   
Monoclinic,  $P2_1/c$   
 $a = 7.1411$  (2) Å  
 $b = 31.4292$  (8) Å  
 $c = 5.7876$  (2) Å  
 $\beta = 91.275$  (2)°  
 $V = 1298.64$  (7) Å<sup>3</sup>

$Z = 4$   
 $D_x = 1.872$  Mg m<sup>-3</sup>  
Mo  $K\alpha$  radiation  
 $\mu = 6.23$  mm<sup>-1</sup>  
 $T = 100.0$  (1) K  
Block, yellow  
 $0.40 \times 0.20 \times 0.12$  mm

#### Data collection

Bruker SMART APEX2 CCD area-detector diffractometer  
 $\omega$  scans  
Absorption correction: multi-scan (SADABS; Bruker, 2005)  
 $T_{\min} = 0.126$ ,  $T_{\max} = 0.530$   
(expected range = 0.113–0.474)

14661 measured reflections  
2962 independent reflections  
2448 reflections with  $I > 2\sigma(I)$   
 $R_{\text{int}} = 0.032$   
 $\theta_{\text{max}} = 27.5^\circ$

#### Refinement

Refinement on  $F^2$   
 $R[F^2 > 2\sigma(F^2)] = 0.047$   
 $wR(F^2) = 0.109$   
 $S = 1.20$   
2962 reflections  
163 parameters  
H-atom parameters constrained

$w = 1/[\sigma^2(F_o^2) + (0.0308P)^2 + 4.6217P]$   
where  $P = (F_o^2 + 2F_c^2)/3$   
 $(\Delta/\sigma)_{\text{max}} < 0.001$   
 $\Delta\rho_{\text{max}} = 1.05$  e Å<sup>-3</sup>  
 $\Delta\rho_{\text{min}} = -0.71$  e Å<sup>-3</sup>

**Table 1**

Hydrogen-bond geometry (Å, °).

<i>D</i> –H··· <i>A</i>	<i>D</i> –H	H··· <i>A</i>	<i>D</i> ··· <i>A</i>	<i>D</i> –H··· <i>A</i>
C2–H2A···Cg2 <sup>i</sup>	0.93	2.75	3.428 (5)	131
C5–H5A···Cg2 <sup>ii</sup>	0.93	2.69	3.342 (5)	127
C14–H14A···Cg1 <sup>iii</sup>	0.93	2.69	3.373 (5)	131

Symmetry codes: (i)  $-x + 2, -y, -z + 2$ ; (ii)  $-x + 1, -y, -z + 1$ ; (iii)  $-x + 1, -y, -z + 2$ . Cg1 and Cg2 are the centroids of the C1–C6 and C10–C15 rings, respectively.

H atoms were placed in calculated positions and refined as riding, with C–H = 0.93 Å and  $U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}(\text{C})$ . The highest peak is located 0.96 Å from atom Br2.

Data collection: APEX2 (Bruker, 2005); cell refinement: APEX2; data reduction: SAINT (Bruker, 2005); program(s) used to solve structure: SHELXTL (Sheldrick, 1998); program(s) used to refine structure: SHELXTL; molecular graphics: SHELXTL; software used to prepare material for publication: SHELXTL, PARST (Nardelli, 1995) and PLATON (Spek, 2003).

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